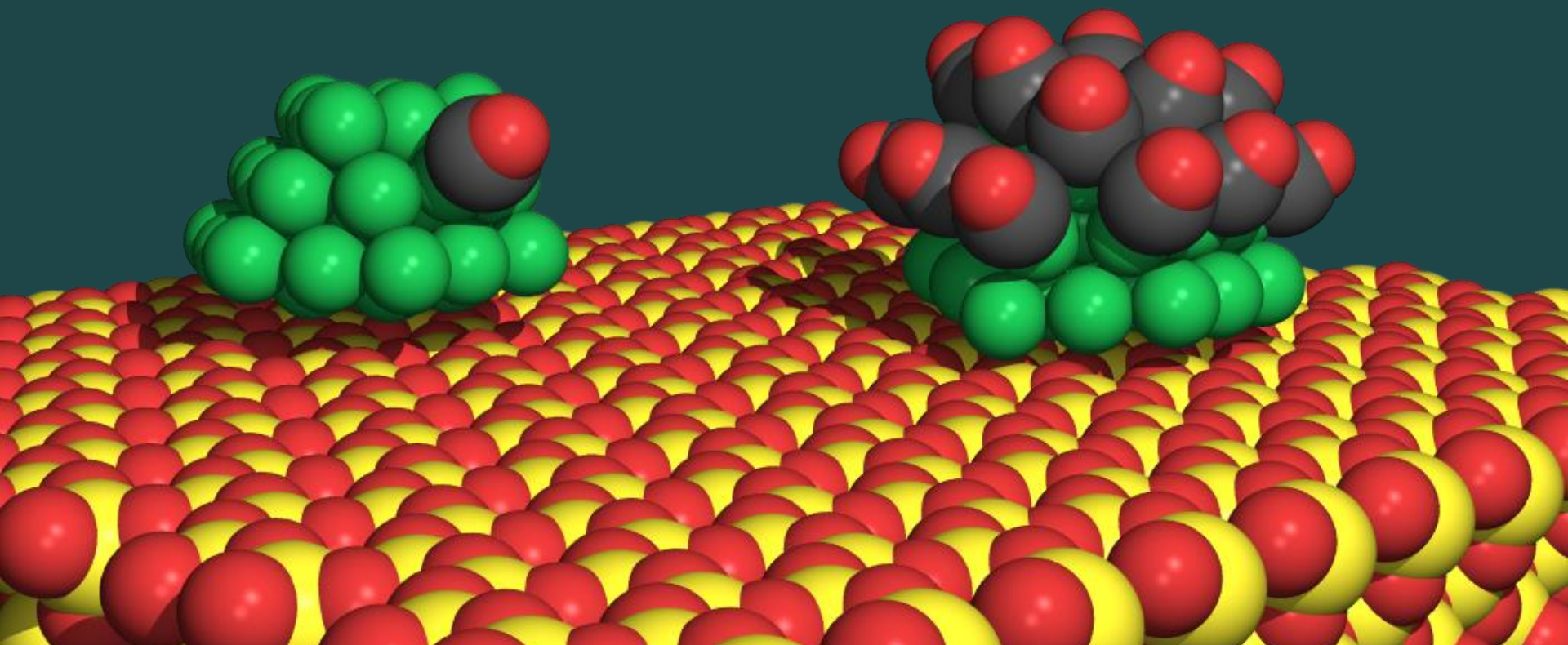


Dynamic structural disorder and reactivity in supported metal nanocatalysts

F.D. Vila, J. J. Rehr and A.I. Frenkel



INDESCRIBABLE...
INDESTRUCTIBLE!
NOTHING CAN STOP IT!

THE BLOB



“A theoretical tour de force”
J. Horr. Phys. A

“Love it..”
An anonymous referee

“Hate it..”
Nature editor

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FERNANDO D. VILA ANATOLY FRENKEL · JOSH KAS

An Evolving Picture of Metal Nanocatalysts

Metal nanocatalysts:

Keystone of **heterogeneous catalysis** in industry

Theoretical studies of nanocatalysts used to:

Use **static** structures

Sample **few** conformations

Not account for **realistic** temperature

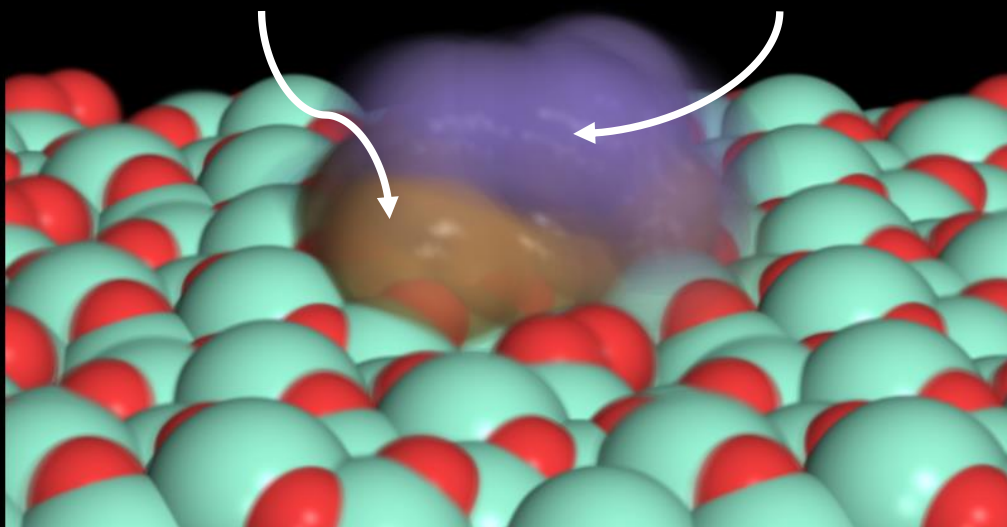
More recently:

Finite temperature DFT/MD simulations

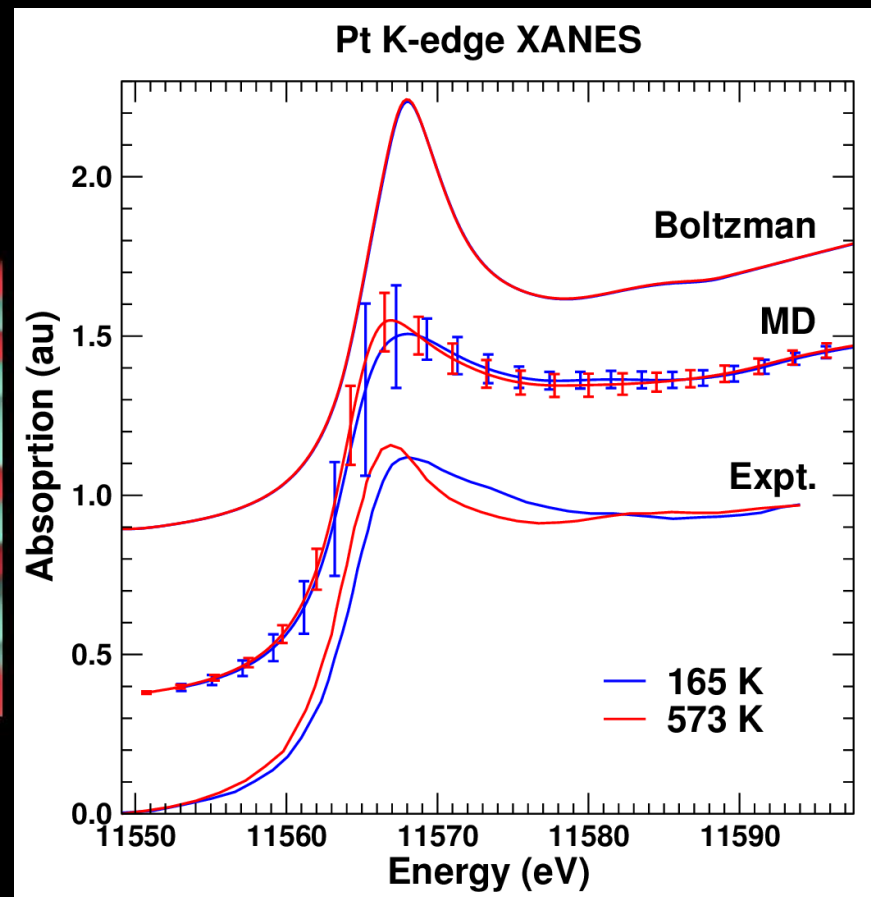
Highlight importance of **disorder**

Static Simulations are Usually Not Sufficient

$+\delta$ ("Oxidized") $-\delta$ ("Metallic")

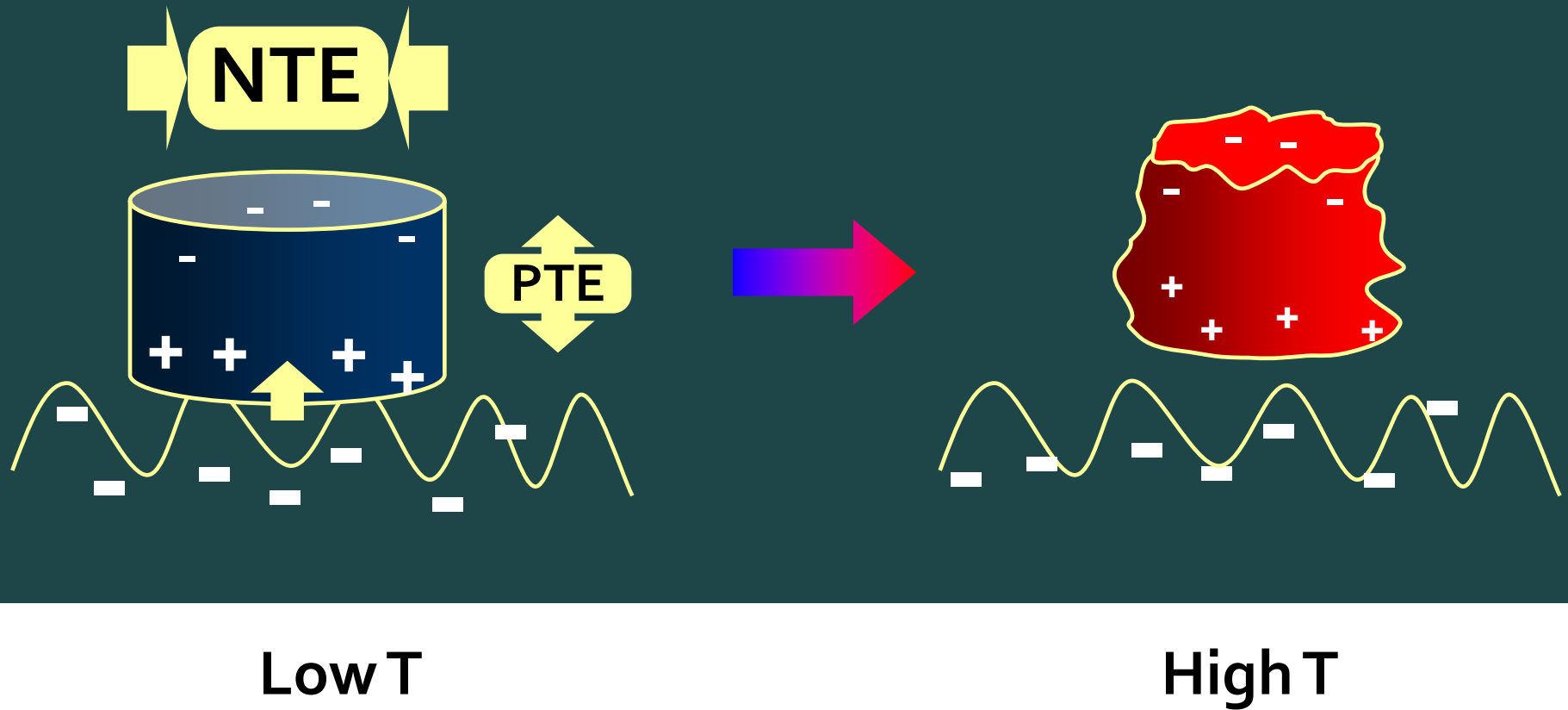


Pt_{10} on $\gamma\text{-Al}_2\text{O}_3$ @ 165 K



Need **dynamics** to reproduce experiment

Model of Negative Thermal Expansion (NTE)



Overall behavior: Results from **inhomogeneous** changes
Driven by **dynamic** fluctuations

Dynamic Structural Disorder (DSD) in Nanoparticles

DSD drives:

Fluctuating bonding

Cluster mobility

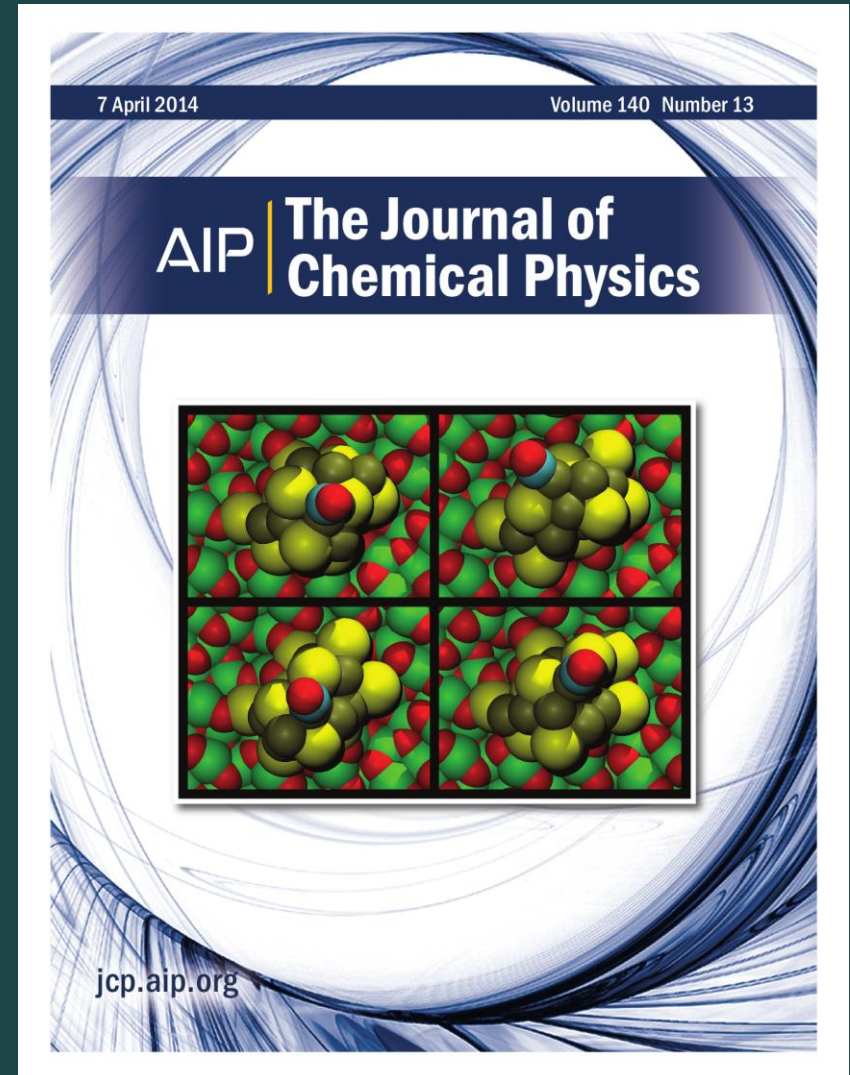
Charge separation

Layering and segregation

Adsorbate dynamics (right)

Adsorbate reactivity

Inhomogeneity



Rehr and Vila J. Chem. Phys. **140**, 134701 (2014)

CO dynamics on Pt₁₀Sn₁₀

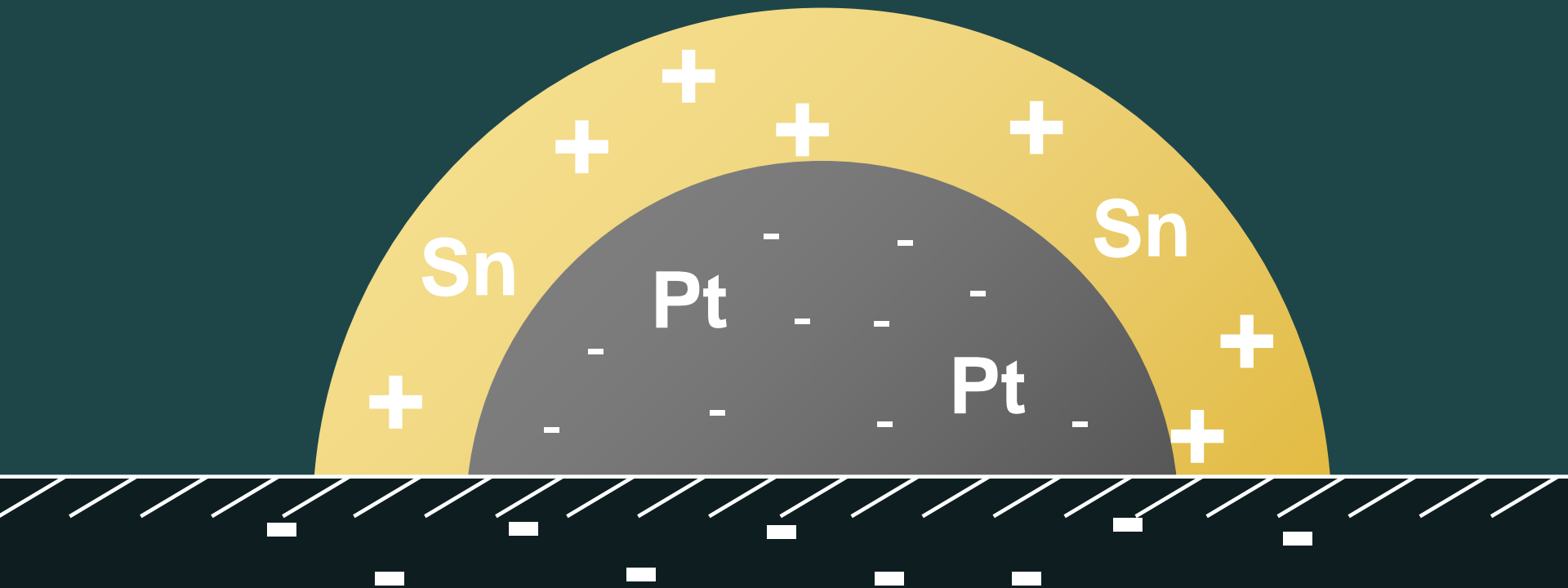
Model of Charge Segregation

In PtSn **alloy** nanoparticles:

Sn **segregation** to particle surface

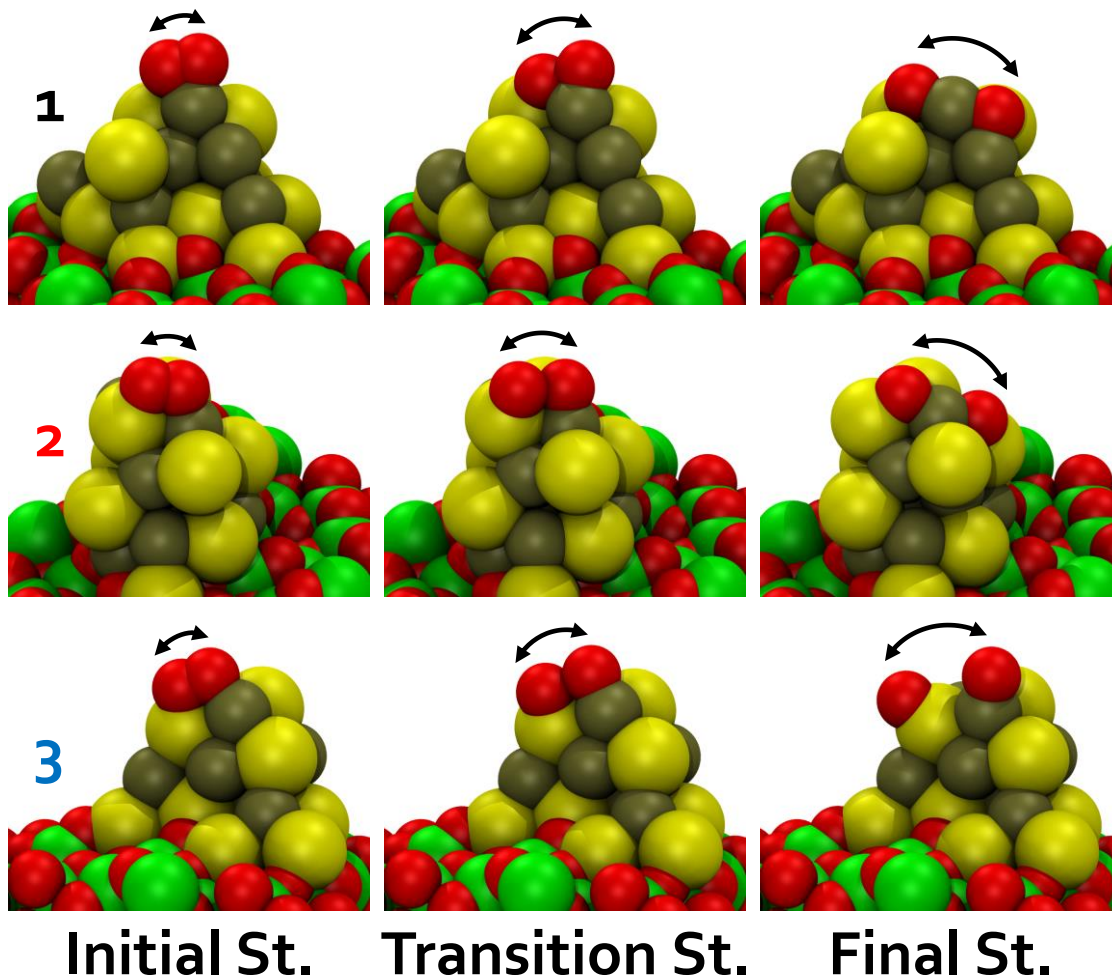
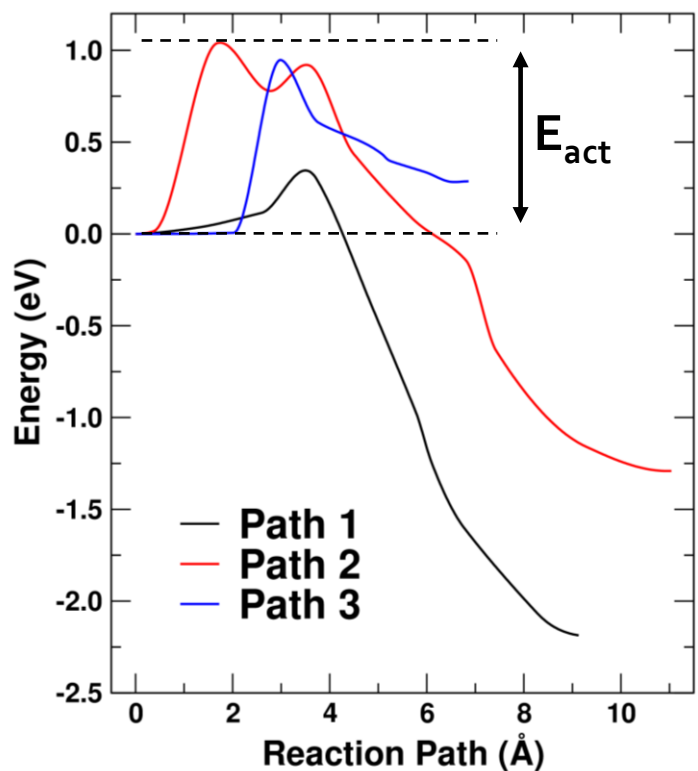
Driven by Coulomb **repulsion**

Less charged **Pt** atoms in **particle core**



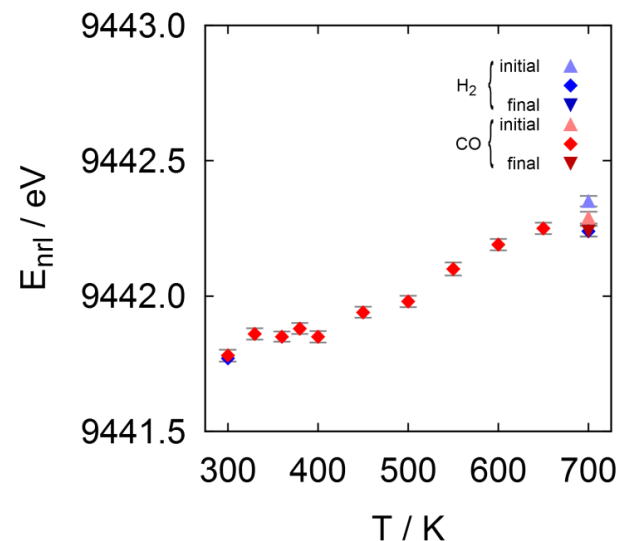
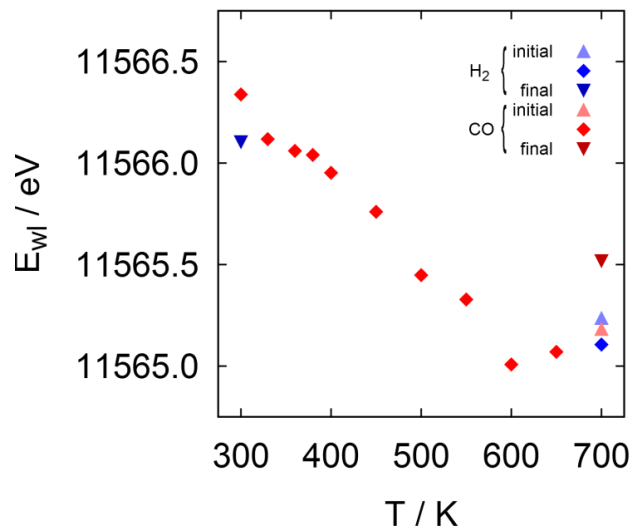
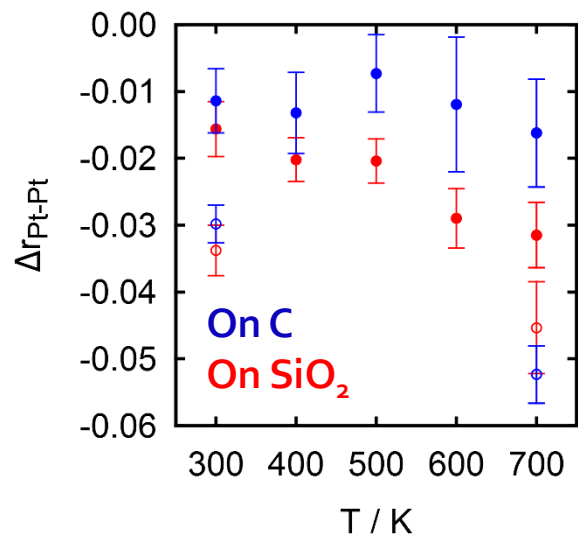
Disorder Affects Reactivity

O₂ dissociation on Pt₁₀Sn₁₀



Large differences in activation energy (E_{act})
Reaction path depends on DSD

Inhomogeneity in Well-defined(?) Nanoparticles



Bond **contraction** with **heating/desorption**

White line: **redshift**, Emission line: **blueshift**

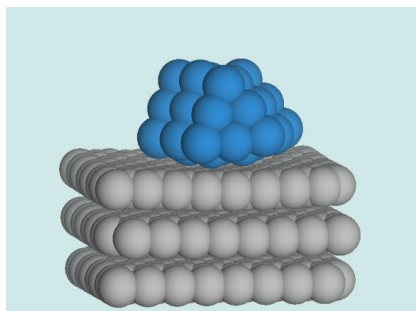
EXAFS measurements: Predict **truncated cuboctahedron** Pt_{37}

Hypothesis: Both phenomena **related** to **desorption**

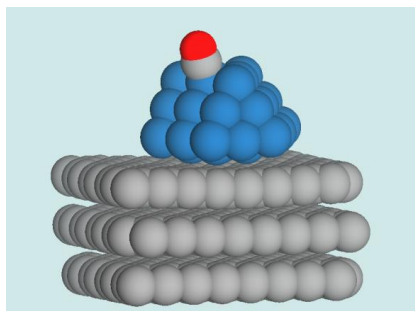
Is **inhomogeneity** important to these phenomena too?

Preliminary Methods and Models

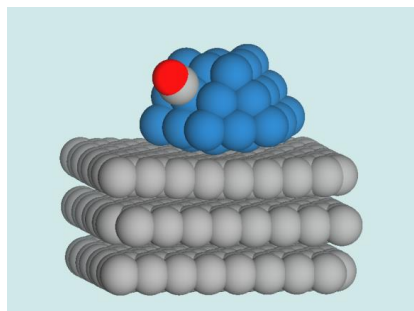
- Pt₃₇ on C and SiO₂:
 - PBE/PAW **optimization** with 400 eV planewave cutoff
 - C surface: 3 **graphite** layers (4 × 4, 384 atoms)
 - SiO₂: **reconstructed** (001) α-quartz (2 × 4, 278 atoms)



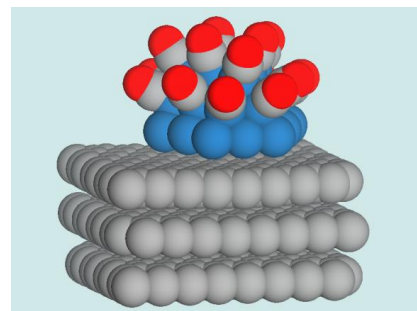
Clean



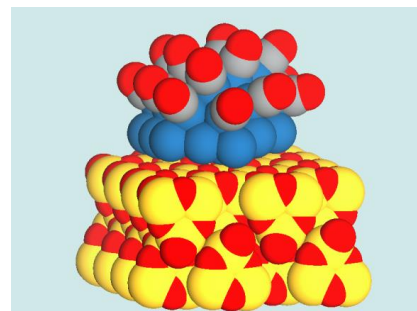
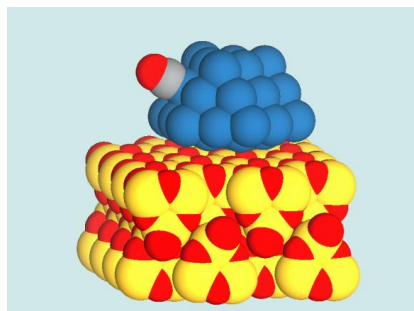
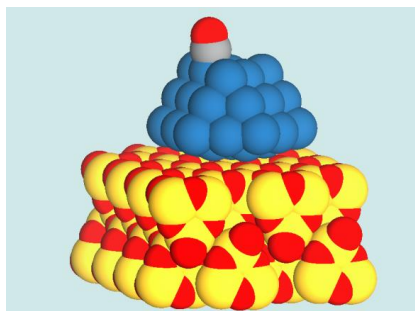
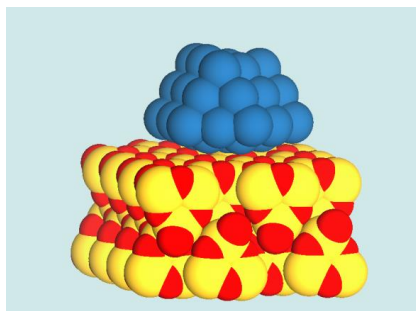
Edge



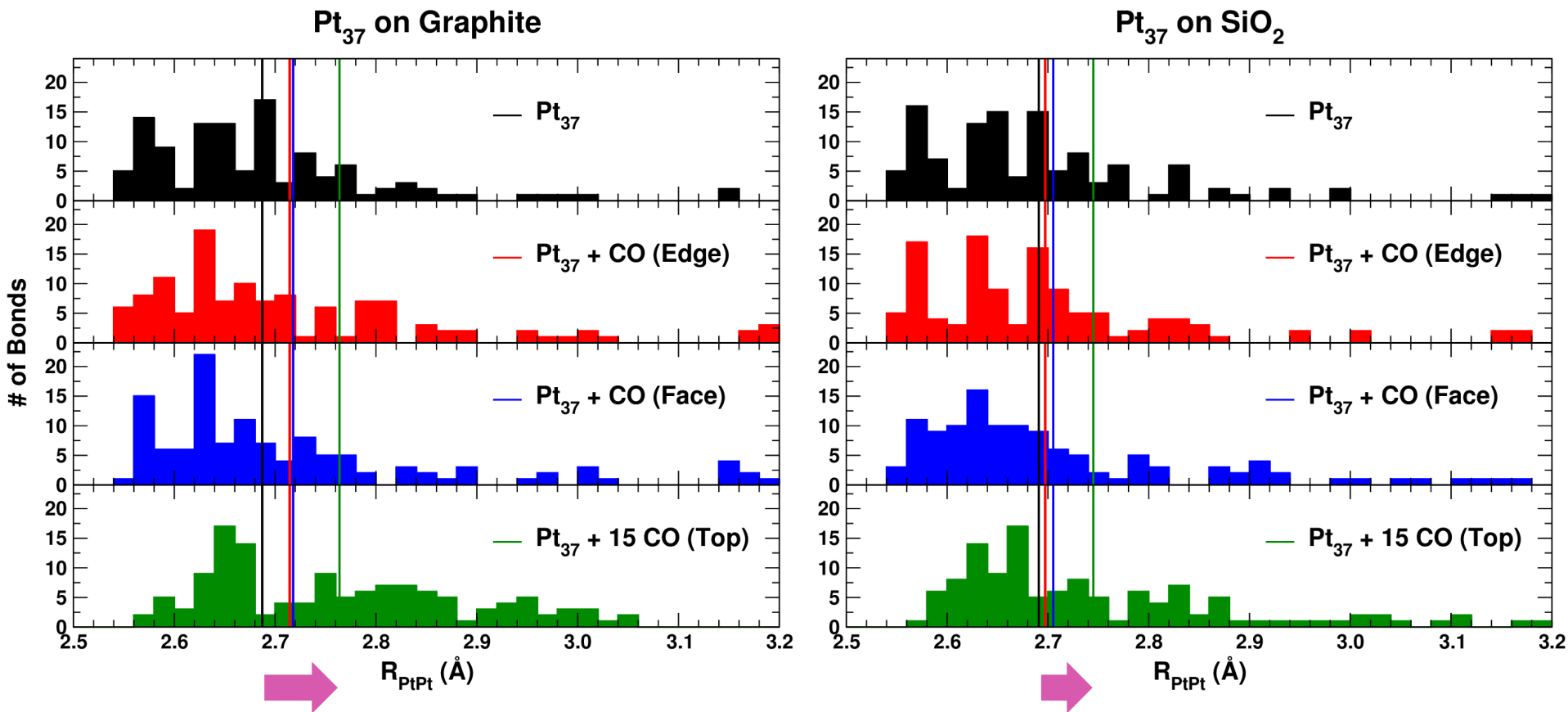
Face



High Cov.

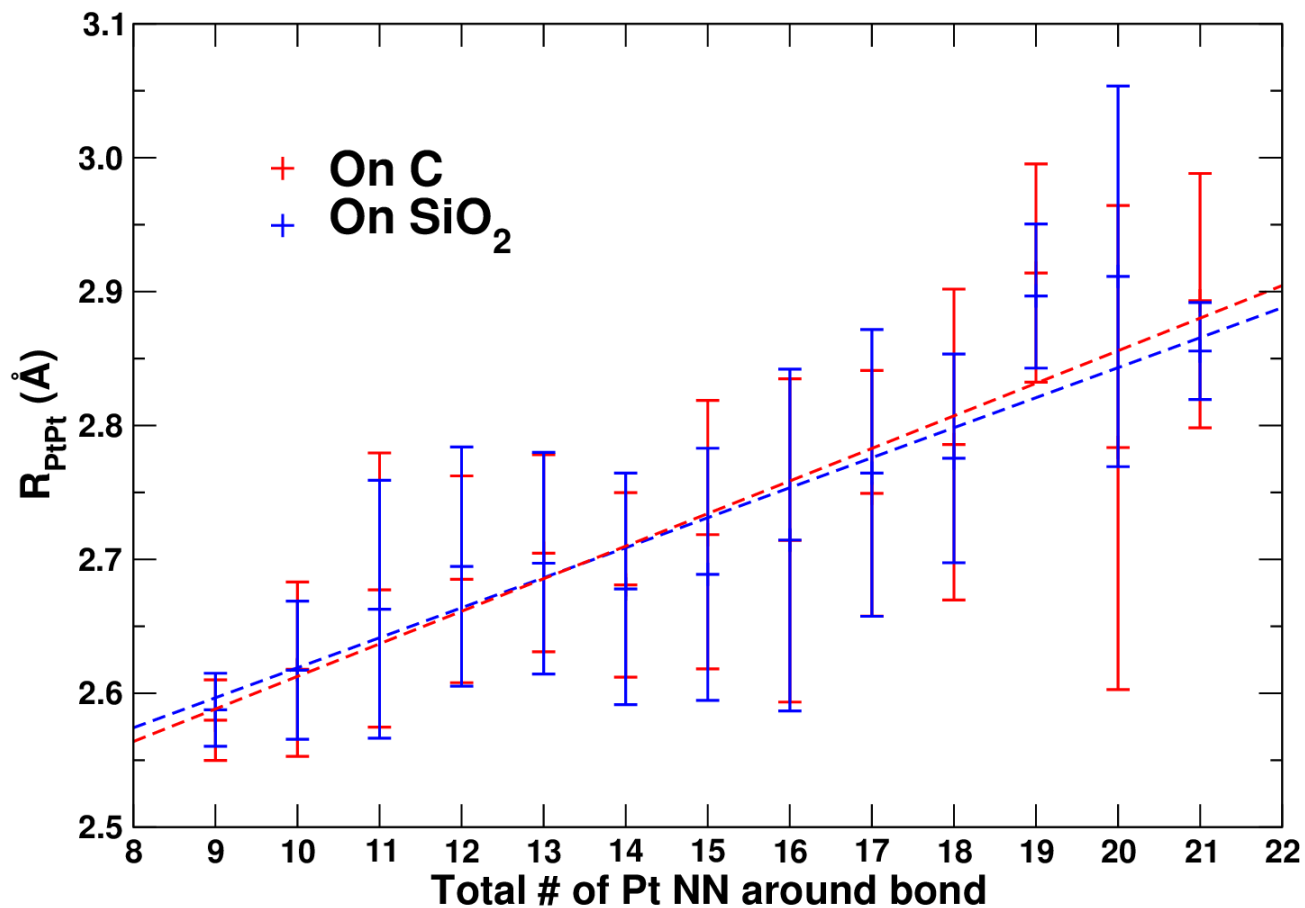


Bond Expansion



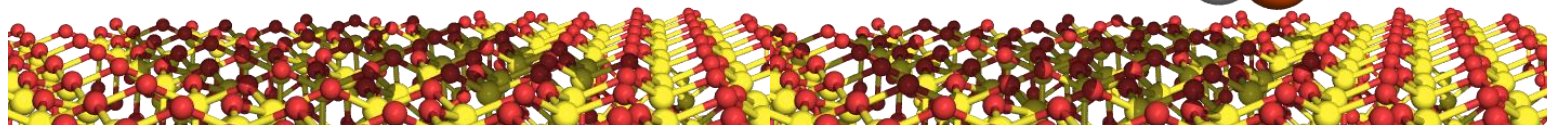
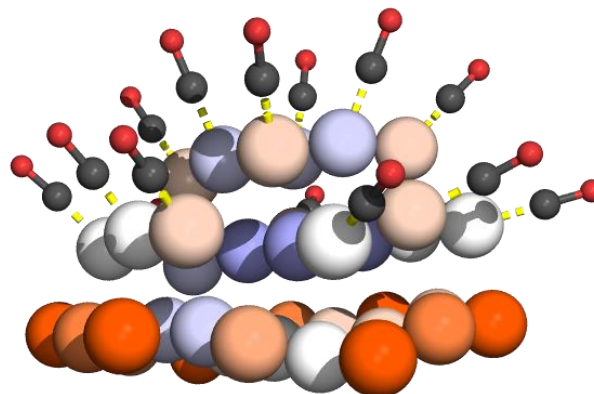
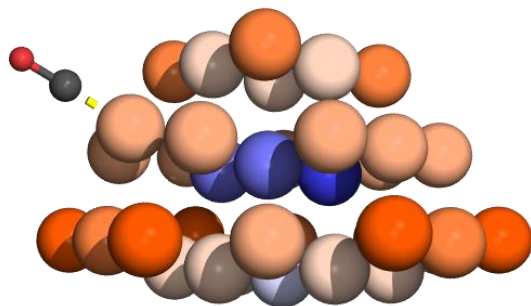
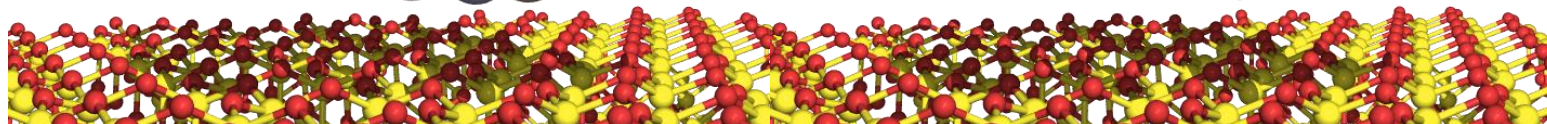
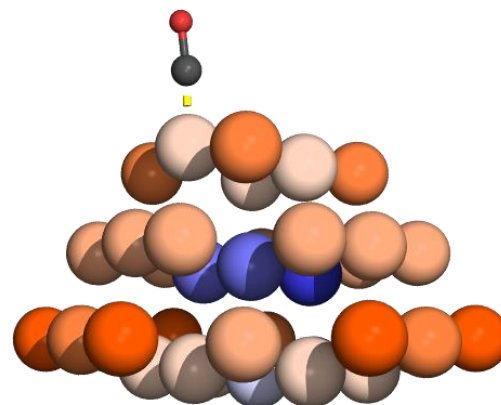
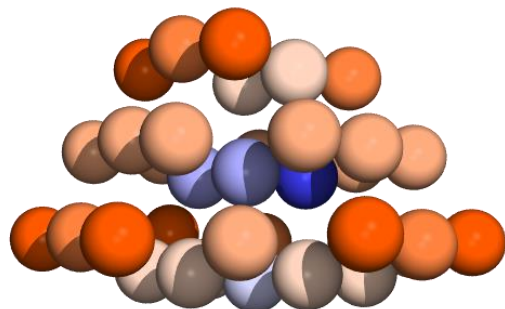
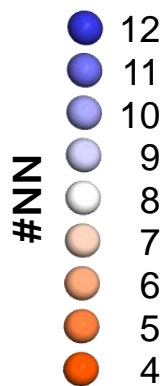
PtPt mean expansion (vs H-covered, not shown):
1.2% on C and 0.4% on SiO₂ (Expt. 1% and 0.4%)

Bond Distance vs # of NN

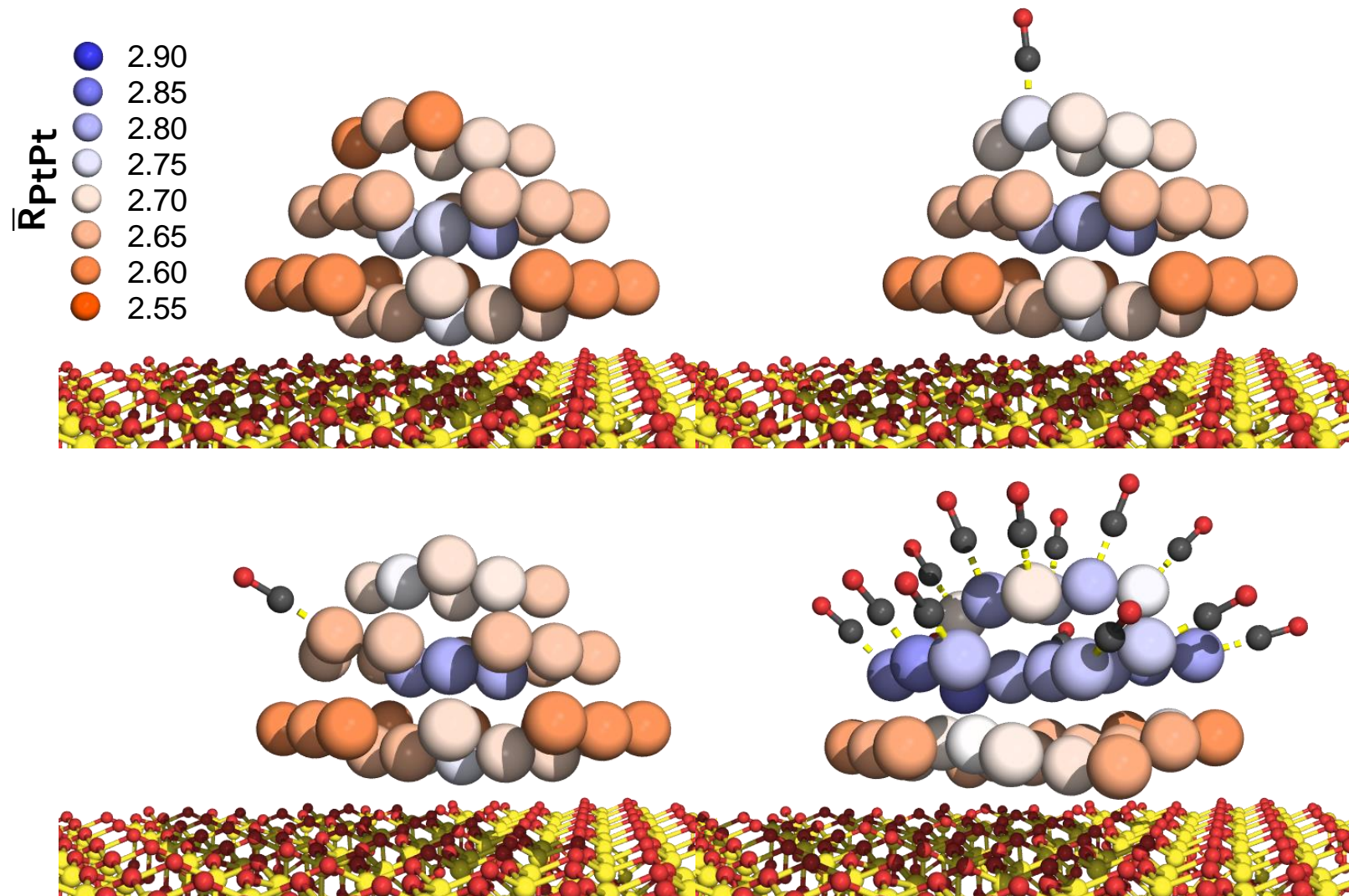


Higher coordination \Leftrightarrow Longer bonds
Similar behavior on both supports

of Pt NN on SiO₂

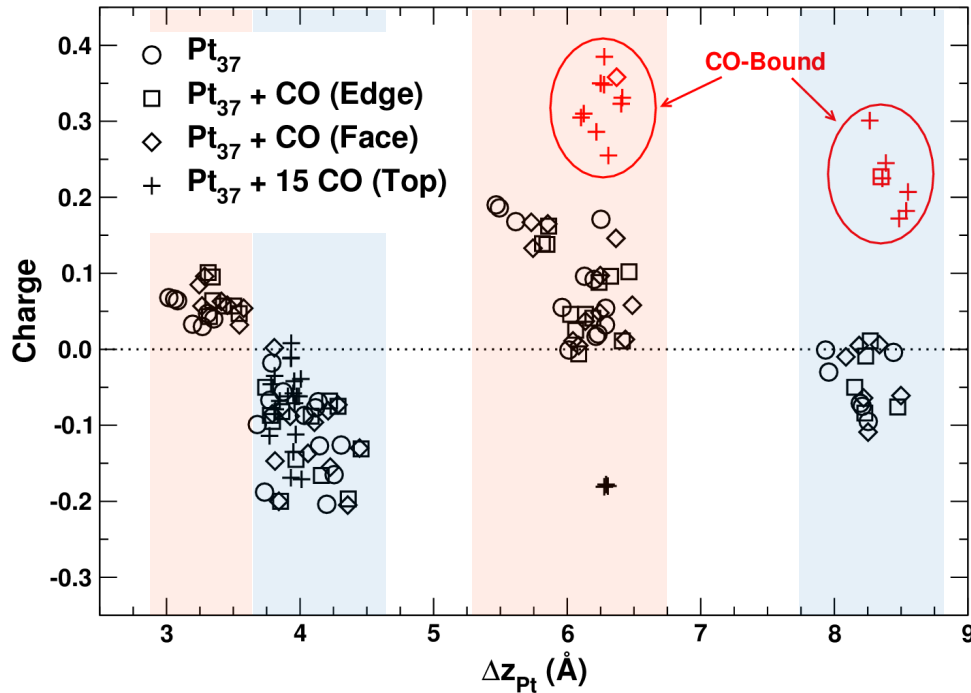


Mean PtPt Bond Lengths on SiO₂

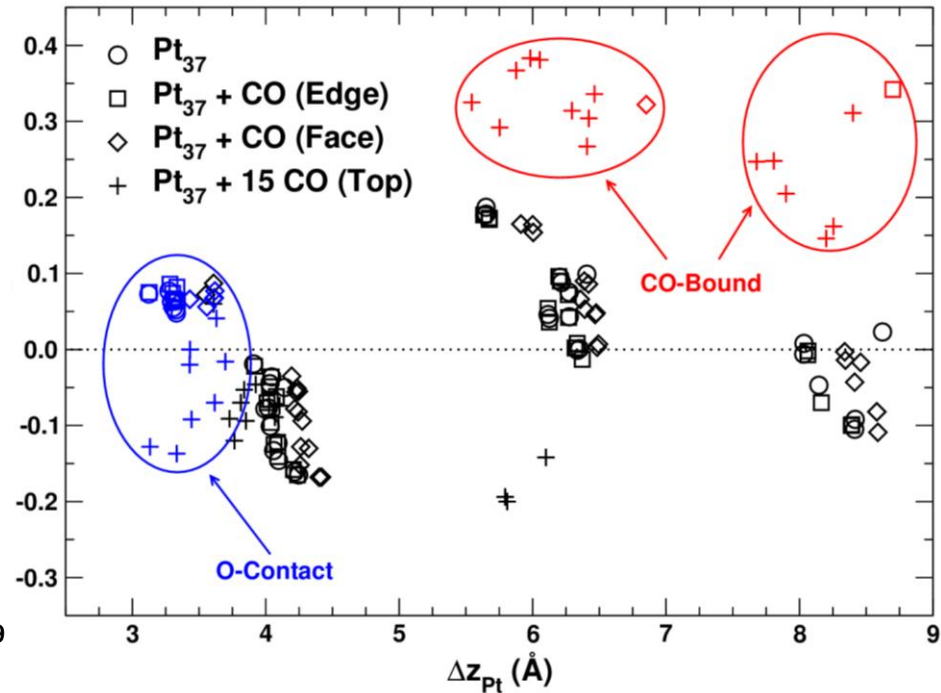


Charge Inhomogeneity

Pt₃₇ on Graphite

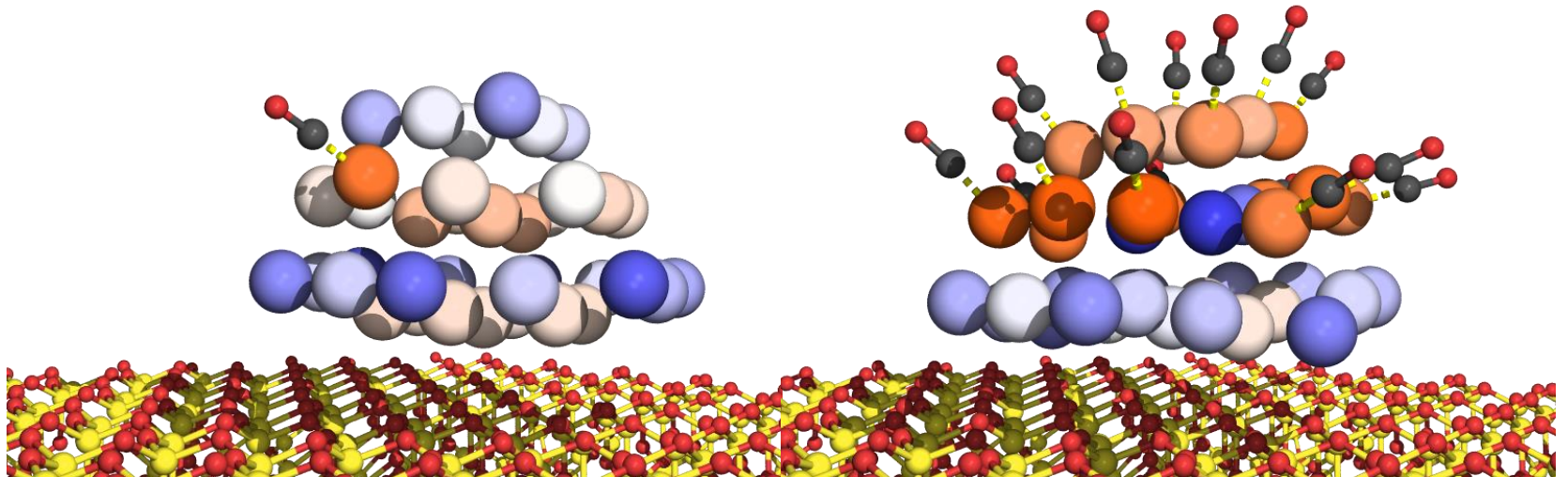
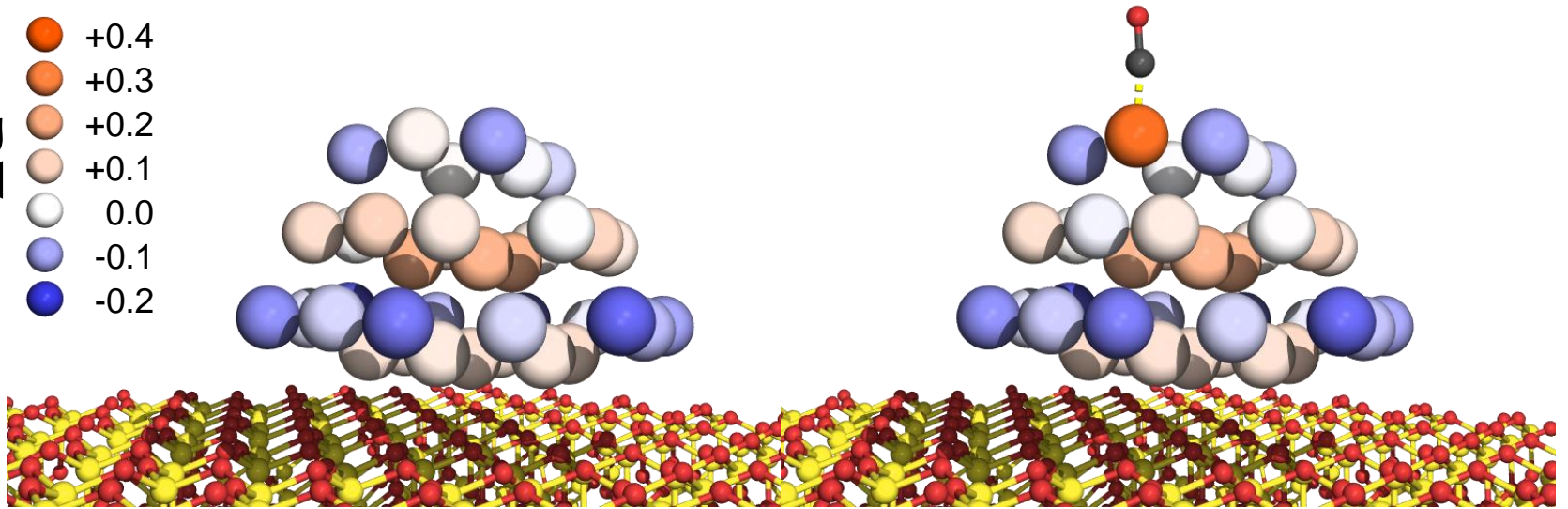
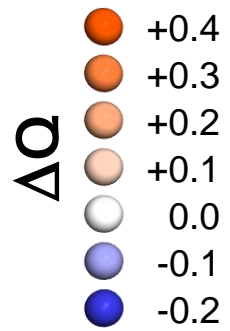


Pt₃₇ on SiO₂



CO-bound Pt atoms **lose 0.2-0.3e** each
Layer **charge alternation**
Bond **expansion** due to **charge loss**

Charge inhomogeneity on SiO₂



Ab initio Vibrational Properties

Projected Vibrational DOS: Use **efficient pole model**

$$\rho_R(\omega) = -\frac{2\omega}{\pi} \operatorname{Im} \left\langle 0 \left| \frac{1}{\omega^2 - \mathbf{D} + i\varepsilon} \right| 0 \right\rangle$$

Dynamical Matrix (D)
from VASP:

$$\cong \sum_{v=1}^N w_v \delta(\omega - \omega_v) \frac{\partial^2 E}{\partial u_{jl\alpha} \partial u_{j'l'\beta}} = (M_j M_{j'})^{-1/2}$$

$$\sigma_R^2(T) = \frac{\hbar}{2\mu_R} \int_0^\infty \frac{1}{\omega} \coth\left(\frac{\beta\hbar\omega}{2}\right) \rho_R(\omega) d\omega$$

Properties: **DW factors**
and mean **Einstein**
temperatures:

$$T_E = \frac{\hbar \langle \omega^2 \rangle^{-1/2}}{k_B} = \frac{\hbar}{k_B} \left(\sum_{v=1}^N \frac{w_v}{\omega_v^2} \right)^{-1/2}$$

Bond Stiffness Inhomogeneity on SiO₂

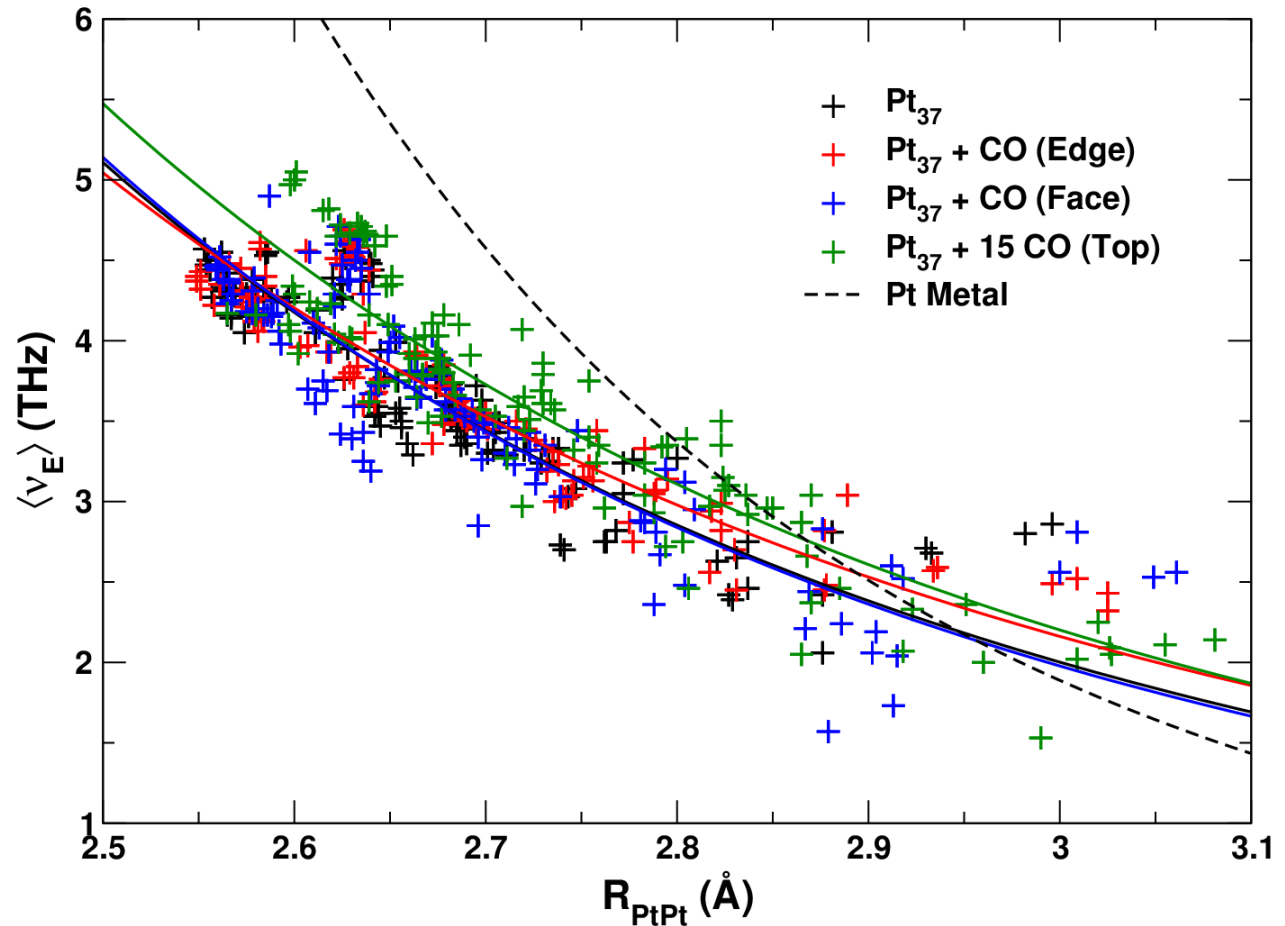
$$\gamma = -\frac{1}{3} \frac{d \ln \langle \nu_E \rangle}{d \ln R_{\text{PtPt}}}$$

Pt metal:

$$\gamma = 2.8$$

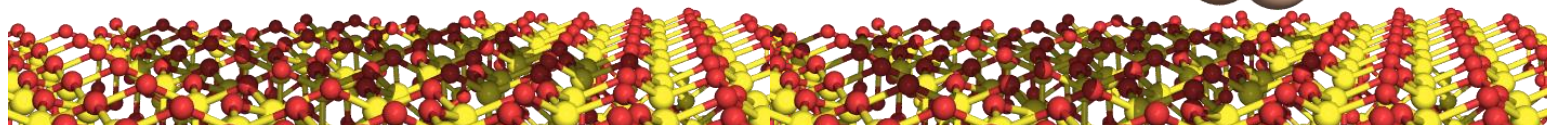
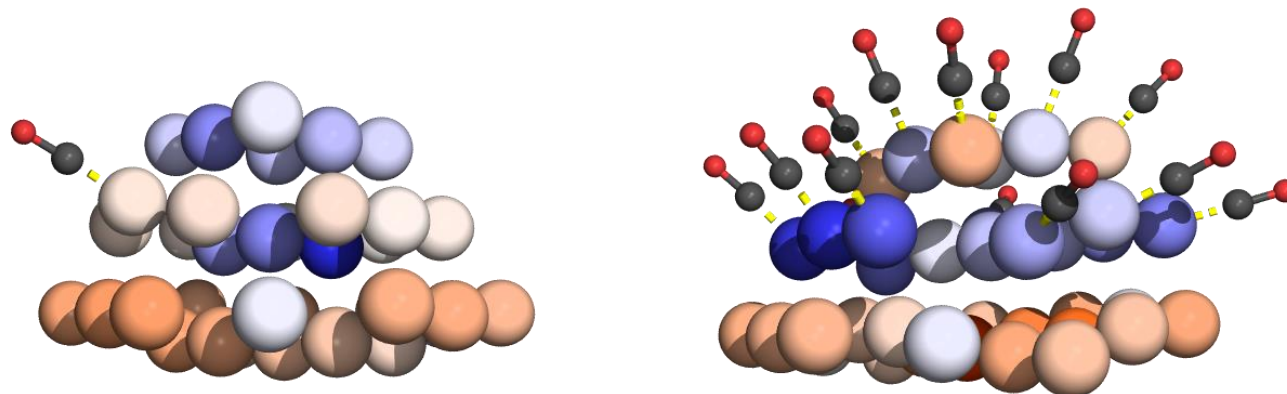
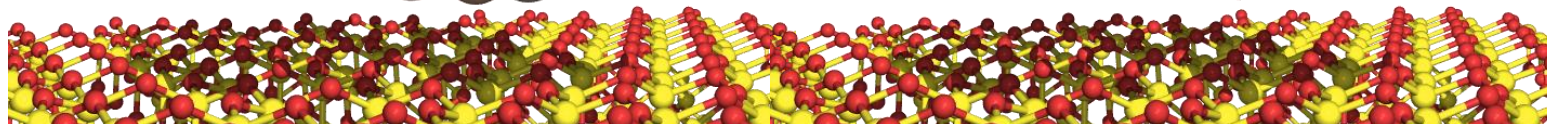
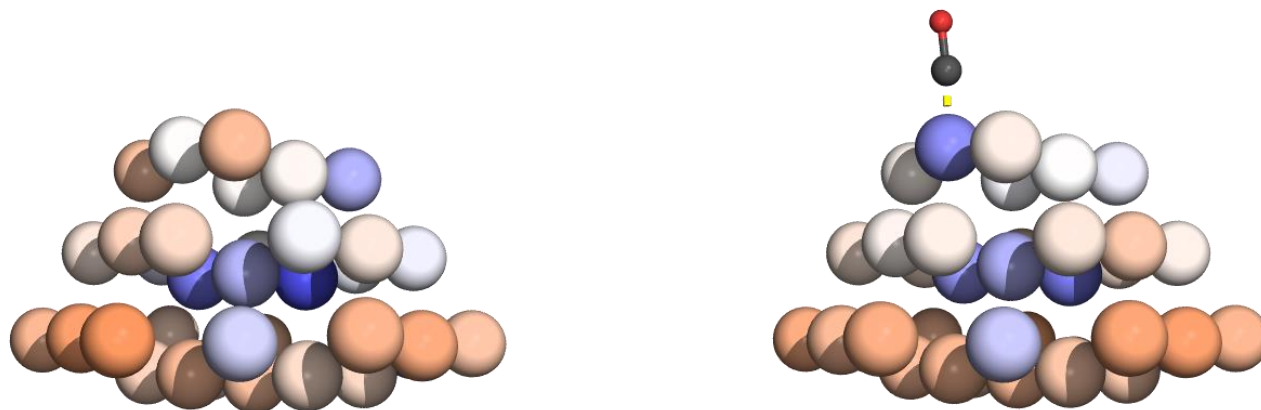
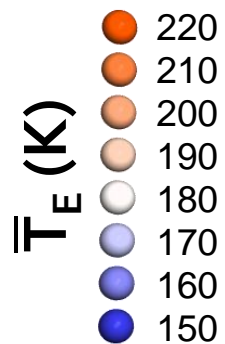
Nanoparticle:

$$\gamma \cong 1.7$$



Broad range of stiffness over single nanoparticle
Reduced Gruneisen parameter (γ) vs metal

Mean PtPt Einstein Temperatures on SiO₂



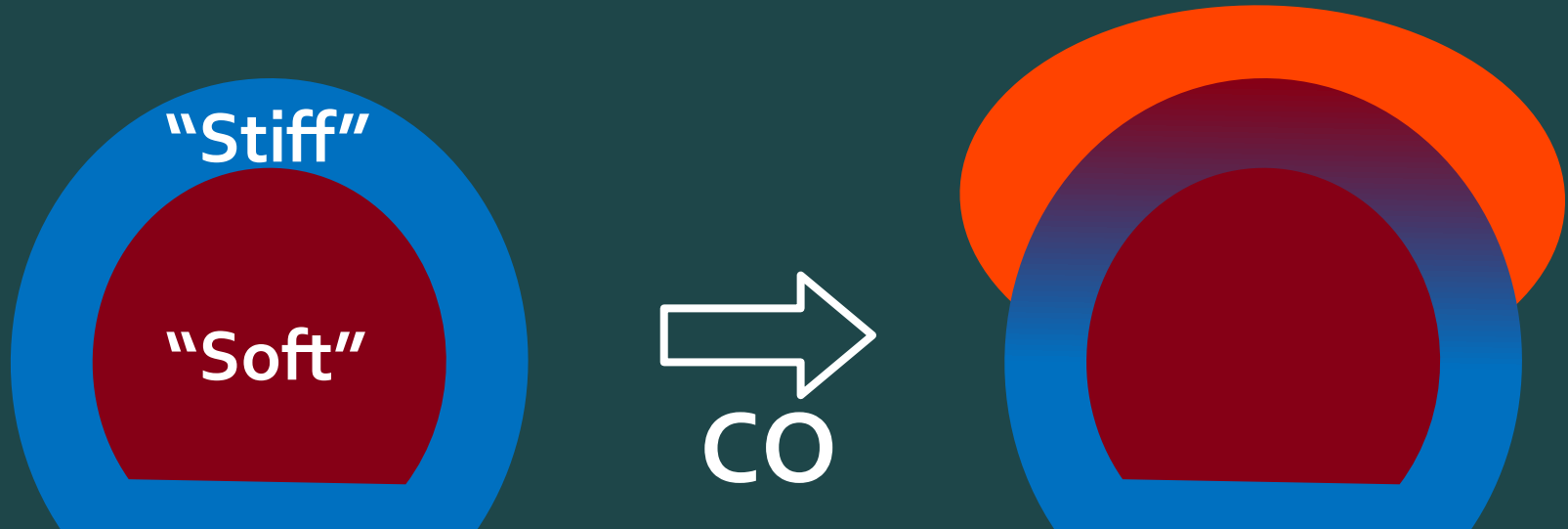
Model of Stiffness

Pt nanoparticles have:

Stiff outer shell (Shorter R_{PtPt} , less #NN)

Soft core (Longer R_{PtPt} , more #NN)

Outer shell weakens upon CO adsorption



Conclusions

- **Inhomogeneity** encompasses nanoparticle behavior:
 - Changes **reactivity**
 - Modulates **charge** distribution
 - Modulates **structure** and **vibrations**
 - **Coupled** to adsorbate interaction
- **Correlations** between XAFS parameters
 - $\#NN \Leftrightarrow R_{PtPt} \Leftrightarrow \nu_E \Leftrightarrow \sigma_{PtPt}^2$
- Future work
 - Finite temperature **dynamics**
 - Local **x-ray spectroscopy**

Structural and charge inhomogeneity in supported Pt clusters

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